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10/575,468	04/12/2006	Hiroyuki Sato	2006_0517A	3215
	7590 06/04/200 , LIND & PONACK, I	EXAMINER		
2033 K STREET N. W. SUITE 800 WASHINGTON, DC 20006-1021			BOYKIN, TERRESSA M	
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			1796	
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)			
	10/575,468	SATO ET AL.			
Office Action Summary	Examiner	Art Unit			
	Terressa M. Boykin	1796			
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address			
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA  - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication.  - If NO period for reply is specified above, the maximum statutory period w.  - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tim vill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	l. lely filed the mailing date of this communication. (35 U.S.C. § 133).			
Status					
Responsive to communication(s) filed on 12 Ag     This action is <b>FINAL</b> . 2b) ☑ This     Since this application is in condition for allowar closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro				
Disposition of Claims					
4)  Claim(s) 1-12 is/are pending in the application.  4a) Of the above claim(s) is/are withdrav  5)  Claim(s) is/are allowed.  6)  Claim(s) 1-12 is/are rejected.  7)  Claim(s) is/are objected to.  8)  Claim(s) are subject to restriction and/or  Application Papers  9)  The specification is objected to by the Examine.  10)  The drawing(s) filed on 12 April 2006 is/are: a)  Applicant may not request that any objection to the or	vn from consideration. relection requirement. r. ⊠ accepted or b)□ objected to l				
Replacement drawing sheet(s) including the correcti		• •			
11) The oath or declaration is objected to by the Examiner. Note the attached Office Action or form PTO-152.					
Priority under 35 U.S.C. § 119					
<ul> <li>12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).</li> <li>a) All b) Some * c) None of:</li> <li>1. Certified copies of the priority documents have been received.</li> <li>2. Certified copies of the priority documents have been received in Application No</li> <li>3. Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).</li> <li>* See the attached detailed Office action for a list of the certified copies not received.</li> </ul>					
Attachment(s)  1) Notice of References Cited (PTO-892)  2) Notice of Draftsperson's Patent Drawing Review (PTO-948)  3) Information Disclosure Statement(s) (PTO/SB/08)  Paper No(s)/Mail Date 11-30-06;4-12-06.	4)  Interview Summary Paper No(s)/Mail Da 5)  Notice of Informal P 6)  Other:	te			

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# Claim Rejections - 35 USC § 102

The following is a quotation of the appropriate paragraphs of 35 U.S.C.

102 that form the basis for the rejections under this section made in this

#### Office action:

(b) the invention was patented or described in a printed publication in this or a foreign country or in public use or on sale in this country, more than one year prior to the date of application for patent in the United States.

Claims 1, 2, 4-8, 9 are rejected under 35 U.S.C. 102(b) as being anticipated by EP 1048683 see pages 1-81 and paragraphs as designated below.

The Examiner first notes that the it is commonly known which will be demonstrated below to produce an aliphatic polyester particularly polyglycolic acid from glycolide

The Examiner also notes that it appears that the crux of the invention in view of the specification is that, in the presence of water the polyglycolide is unstable. it appears to the examiner that applicants have attempted to stabilize the polyglycolide by adding end capping agents such as those claimed affording a more stable polymer event under hydrolytic conditions.

The instability of the polyester is well known and addressed in the art and the positive

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aspects due to its degradability and thus useful in the biomedical field of the polyester is also well known.

Thus, with regard to applicants' claims 1 and 2 which is directed to a process for producing an aliphatic polyester, comprising: subjecting a cyclic ester containing water in excess of 80 ppm as an initiator or/and a molecular weight-adjusting agent to ring-opening polymerization based on a total proton concentration in the cyclic ester as an index, and compounding a resultant aliphatic polyester with a carboxyl group-capping agent which may be an epoxy compound.

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The reference **EP 1048683** addresses the problem and/or advantage of biodegradation like applicants by adding an agent such as those disclosed. For example in **EP1048683** discloses aliphatic polyesters which are composed of at least one kind of monomer unite of any one of a lactone or lactide composition. [0069]. The aliphatic polyester contains a bond as shown above which allows reaction between it and a modifying agent such as a n alcohol terminal moiety or a carboxylic acid terminal moiety.

Note for example in paragraph [00187] that a monoepoxy compound may be employed fro capping the terminals of the polymer chain.

Note also for example in [00468] that diethylene glycol was used as an initiator with octylate catalyst are added to caprolactone monomer having a water content of 150ppm and an acid value of .10 mg-KOH/g and a lactide monomer having a water content of 70ppm and an acid value of .12 mg-KOH/g to prepare a caprolactone/lactide block copolymer, a proportion of the moiety has an alcohol terminal and a carboxylic terminal of 63% and 37% respectively. the lactide was polymerized after polymerization of caprolactone. Note also that catalyst are employed that may be metal compounds

such as Al, Ti, Sn and Zn including metal oxides although not limited thereto.

With regard to claims 10, 11, and 12 note that the reference discloses that the cyclic ester monomer can be polymerized by ring opening through continuously supply the cyclic ester monomer in to a continuous melt-polymerization apparatus. The apparatus may be a column type reactor and may be employed in combination of two or more kinds which would anticipate applicants "plurality of tubes". See [0200] –[0202]

Turning now to **EP 0299730** note also pages 1-2 and claims of **EP 0299730** disclose a lactide and a glycolide which are copolymerized to prepare a dilactic acid glycolic acid copolymer. the polymerization is performed in the presence of a regulator which is alpha hydroxy lower fatty acid containing water and in the amount of .25 to 50wt. % based on the sum of the lactide and the glycolide. The catalyst is stannous octoate. The process consist of copolymerizing lactide and glycolide using a regulator which may be alpha hydroxy lower fatty acid such as glycolic acid and water. The amounts as disclosed in the abstract and in claims appear anticipate those amounts as disclosed. Note that the reference discloses the use of a polymerization regulator which may anticipate the molecular weight adjuster as claimed by applicants.

Thus, each of the references **EP 1048683 or EP 0299730** discloses a ring-opening polymerization reaction to produce a polyester prepared from the same components as claimed by applicants. Since the disclosed amounts of water in ppm are expressed differently and thus may be distinct from those claimed, it is incumbent upon applicant(s) to establish that they are in fact different and whether such difference is unobvious. In view of the above, there appears to be no significant difference between

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the reference(s) and that which is claimed by applicant(s). Any differences not specifically mentioned appear to be conventional. Consequently, the claimed invention cannot be deemed as novel and accordingly is unpatentable.

## Claim Rejections -35 USC 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis or all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claim 2 is rejected under 35 U.S.C. 103(a) as being unpatentable over either EP 1048683 or EP 0299730 in view of USP 5885709 or 2937164 see abstract, claims.

With regard to applicants' claims 2 and 3 wherein the carboxyl group-capping agent is selected from the group consisting of monocarbodiimide, polycarbodiimides, oxazolines, and oxazines and epoxy compounds.

Both the **EP 1048683 and the EP 0299730** reference discloses a ring opening polymerization of an aliphatic polyester by the process as claimed except for specifying the particular end capping agent as claimed other than an epoxy moiety. The end capping agents listed in claim 2 are well-known in the art.

For example, **USP** 5885709 discloses a polyester which may be reacted with carbodiimides, as capped carboxyl end groups, the carboxyl end groups being predominantly capped by reaction with mono- and/or biscarbodiimides which are present in the fibers and filaments in an amount of from 30 to 200 ppm, based on the weight of the polyester, the free carboxyl end group content being less than 3 meq/kg of polyester, and the fibers and filaments additionally containing at least 0.02 percent by weight of at least one free polycarbodiimide or of a reaction product containing still reactive carbodiimide groups, and also a process for the preparation thereof.

Note also **USP 2937164** discloses cross-linked polymer compositions and methods for their preparation and more specifically to cross-linked products resulting from the reaction between linear organic acid containing polymers :and monocarbodiimides and the method of preparing the 20 cross-linked polymers. The term "monocarbodiiinide" as used herein includes compounds having one carbodiimide group corresponding to the general formula R-N=C--N-R' in which R and R' are alkyl, substituted alkyl, alkenyl, aryl, aralkyl, 25 alkyl aryl and substituted aryl.

Thus, it would have been obvious to one having ordinary skill in the art at the time the invention was made to employ the particular end capping agent as disclosed in either USP 2937164 or USP 5885709 since both reference discloses the benefits of employing the particular agent as a economically useful and rapidly processable reaction. One would have been motivated to use the end capping agent since both EP 1048683 or EP 0299730 discloses that the end capping agent affords a more stable and regulatable polyester.

Claim 3 is rejected under 35 U.S.C. 103(a) as being unpatentable over both EP 1048683 or EP 0299730 see abstract, claims in view of JP 57-094019.

Both the **EP 1048683 and the EP 0299730** reference discloses a ring opening polymerization of an aliphatic polyester by the process as claimed except for specifying the particular end capping agent as claimed other than an epoxy moiety. The end capping agent as claimed in applicants' claim 3 is well-known in the art.

JP 57-094019 discloses a polyester-polyamide having excellent compatibility with resins or solvents, by reacting a bis(2-oxazoline) compound with a polycarboxylic acid at a specified ratio. Specifically the reference discloses a bis(2-oxazoline) compound (A), e.g., 2,2'-(1,3-phenylene)bis(2- oxazoline), reacted with a polycarboxylic acid (B), e.g., adipic acid or trimellitic acid, carboxyl groups of component B per mol component A (in the presence or absence of a solvent at a reaction temperature≥about 90°C). note that a carboxylic acid is in fact a carbonic acid ester or a polyester.

Thus, it would have been obvious to one having ordinary skill in the art at the time the invention was made to employ the oxazoline compound with the polycarboxylci acid as noted above since the reference discloses that such may be used as an end capping agent since the reference discloses the benefits of employing the particular agent as a useful in the reaction. One would have been motivated to use the end capping agent since both **EP 1048683 or EP 0299730** discloses that the end capping agent affords a more stable and regulatable polyester.

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# Claim Rejections - 35 USC § 112

The following is a quotation of the second paragraph of 35 U.S.C. 112:

The specification shall conclude with one or more claims particularly pointing out and distinctly claiming the subject matter which the applicant regards as his invention.

Claim 8 is rejected under 35 U.S.C. 112, second paragraph, as being indefinite for failing to particularly point out and distinctly claim the subject matter which applicant regards as the invention.

Note that the recited "cap<u>able</u>" is not e definitive as to whether the ring opening is necessary to occur for the polymerization to take place or is applicant merely stating intent that the cyclic monomer possess this ability if needed for future processing.

### Provisional Obviousness-type Double Patenting

The nonstatutory double patenting rejection is based on a judicially created doctrine grounded in public policy (a policy reflected in the statute) so as to prevent the unjustified or improper timewise extension of the "right to exclude" granted by a patent and to prevent possible harassment by multiple assignees. A nonstatutory obviousness-type double patenting rejection is appropriate where the conflicting claims are not identical, but at least one examined application claim is not patentably distinct from the reference claim(s) because the examined application claim is either anticipated by, or would have been obvious over, the reference claim(s). See, e.g., *In re Berg*, 140 F.3d 1428, 46 USPQ2d 1226 (Fed. Cir. 1998); *In re Goodman*, 11 F.3d 1046, 29 USPQ2d 2010 (Fed. Cir. 1993); *In re Longi*, 759 F.2d 887, 225 USPQ 645 (Fed. Cir. 1985); *In re Van Ornum*, 686 F.2d 937, 214 USPQ 761 (CCPA 1982); *In re Vogel*, 422 F.2d 438, 164 USPQ 619 (CCPA 1970); and *In re Thorington*, 418 F.2d 528, 163 USPQ 644 (CCPA 1969).

A timely filed terminal disclaimer in compliance with 37 CFR 1.321(c) or 1.321(d) may be used to overcome an actual or provisional rejection based on a nonstatutory double patenting ground provided the conflicting application or patent either is shown to be commonly owned with this application, or claims an invention made as a result of activities undertaken within the scope of a joint research agreement.

Effective January 1, 1994, a registered attorney or agent of record may sign a terminal disclaimer. A terminal disclaimer signed by the assignee must fully comply with 37 CFR 3.73(b).

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Claims 1-12 are provisionally rejected under the judicially created doctrine of obviousness-type double patenting over claims of copending Application No.

10/577379. Note that the claims of the application although stated differently convey and contain therein the same limitations as that of the present application. note claim 1 for examples also discloses a process for producing an aliphatic polyesters by subjecting the cyclic ester containing water and an initiator, the application is specific with regard to the initiator in stating that it is an alcohol, the claims also stress that a molecular weight adjusting agent to ring opening polymerization based on the total

This is a provisional obviousness-type double patenting rejection since the conflicting claims have not yet been patented.

#### Correspondence

proton concentration is claimed.

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Terressa M. Boykin whose telephone number is 571 272-1069. The examiner can normally be reached on Monday-Thursday 10-5:30 Friday (work at home).

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If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, James Seidleck can be reached on 571 272-1078. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

/Terressa M. Boykin/ Primary Examiner, Art Unit 1796